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Single-Shell Tank Waste Stability

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Single-Shell Tank (SST) Waste Stability

Agenda

- Processes that generated chemicals contained in Hanford Site SSTs
- Major chemicals of concern
- Potential reactions of concern
- Studies to evaluate explosion potential
- Kyshtym explosion
- Relevance of Kyshtym to Hanford Site SST Waste
- Summary and Conclusions

Hanford Site Processes

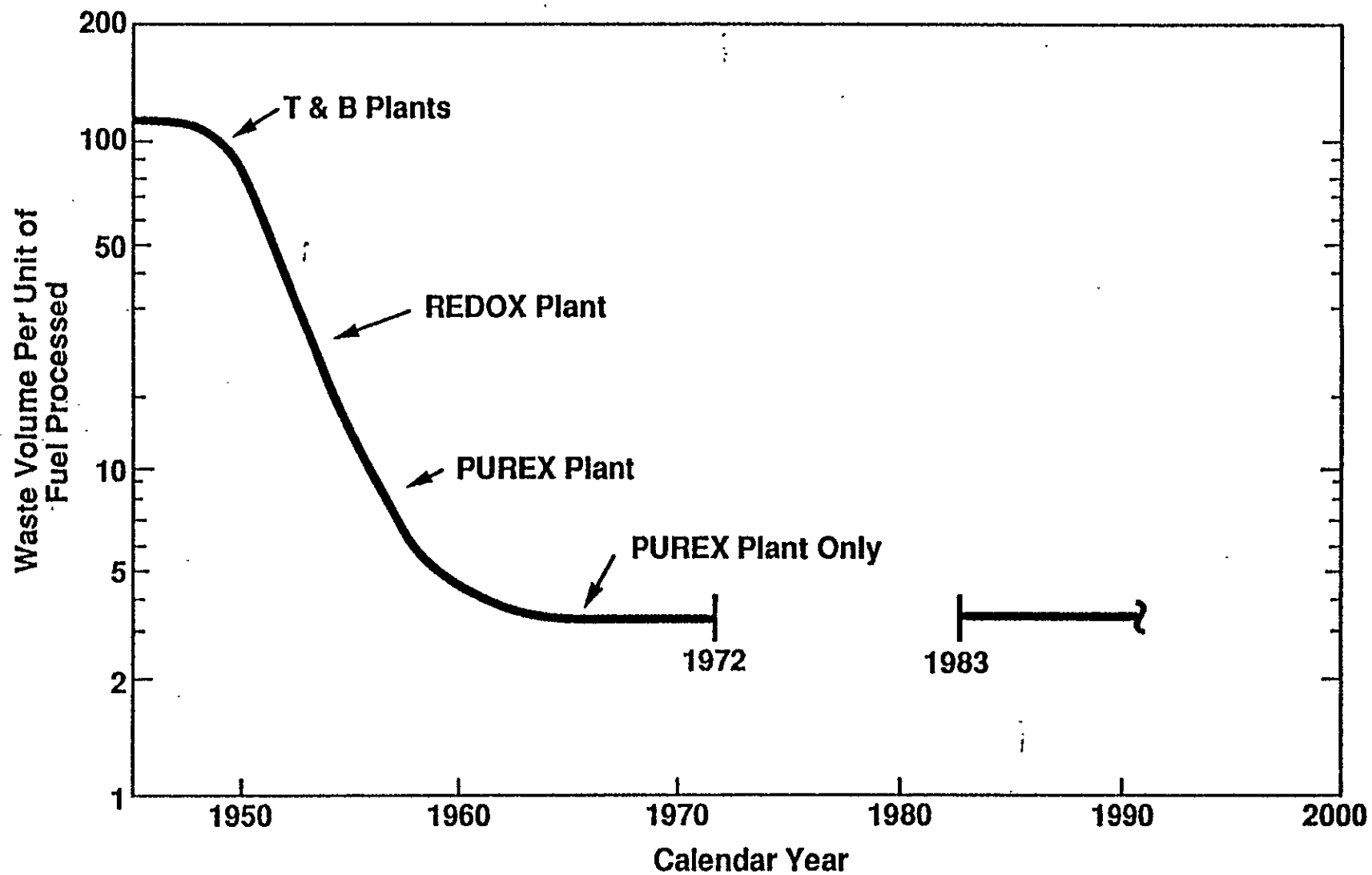
Primary Recovery Process

Process	MTHM* Total Fuel Processed	Operating Period				
		1940s	1950s	1960s	1970s	1980s
Bismuth Phosphate	7,000	12/44	2/56			
REDOX	19,000		1/52	8/66		
PUREX	75,000		1/56	12/71		11/83
Total	101,000					

*Metric tons of heavy metal

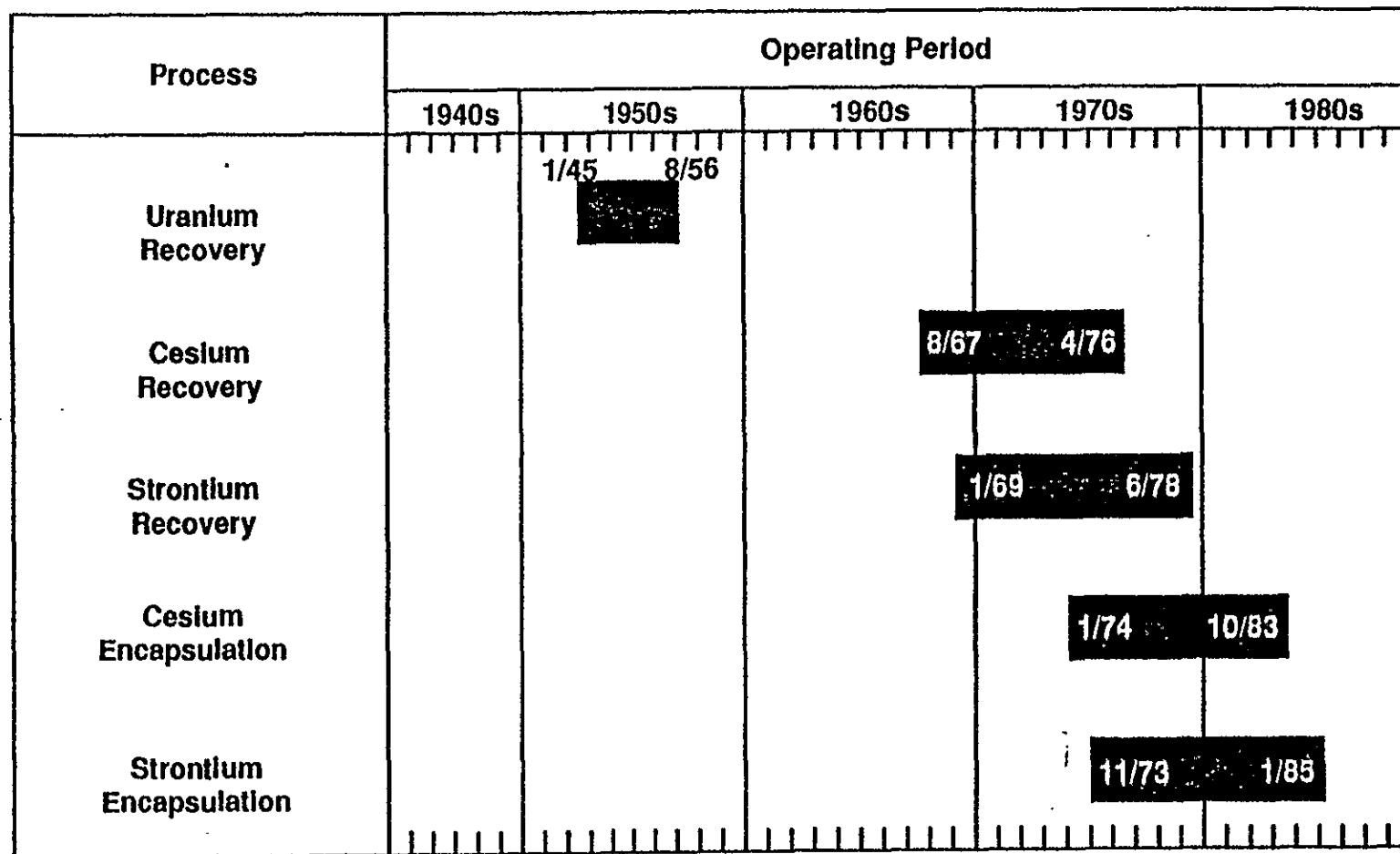
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Hanford Site Processes Waste Generator



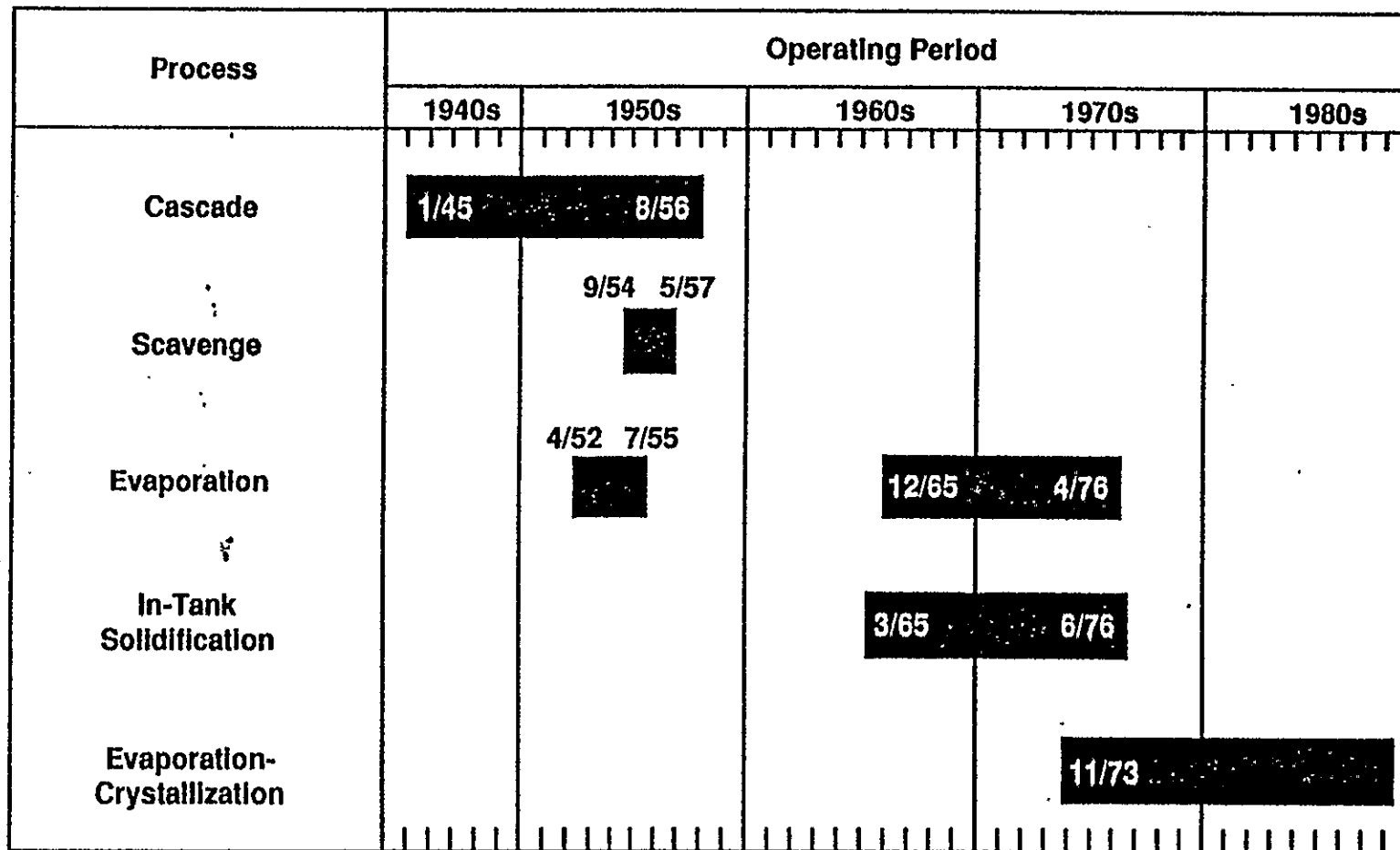
Hanford Site Processes

Secondary Recovery Process



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Hanford Site Processes Waste Management Process



Major Chemicals of Concern

Bismuth Phosphate, REDOX, and PUREX

Classification

Ammonium Nitrate

Explosive

Organics Used
(and Breakdown Products)

Reductants

Inorganic Nitrates and Nitrites

Oxidizers

Major Chemicals of Concern (cont)

Other Sources: Uranium and plutonium recovery, N Reactor operations, laboratories, waste fractionation and encapsulation, equipment decontamination, and waste scavenging

- **Chemicals of concern--same as previous processes but in smaller amounts**
- **Exception--large amount of ferrocyanide (FeCN) used during scavenging process (22 tanks)**

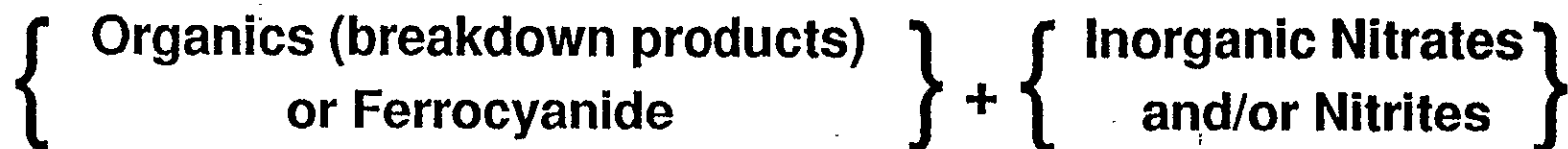
Potential Reactions of Concern

- Ammonium nitrate explosion

All ammonium nitrate in waste destroyed because



- Reduction-oxidation reactions



Studies to Evaluate Explosion Potential

Event	Year
• Kyshtym Accident	1957
• Kyshtym Accident First Revealed	1973
• Nitrate and Organics Used (Rockwell Hanford Operations/ Hazards Research Institute [HRI])	1973-1977
• Literature Review of FeCN Potential (Pacific Northwest Laboratory [PNL])	1983-1984
• Literature Review of Organics Potential (PNL)	1985
• Laboratory Test of FeCN Potential (PNL)	1988-1990
• Kyshtym Accident Officially Confirmed	1989
• Evaluation of Kyshtym Event, and Relevance to Hanford Site Tanks (Westinghouse Hanford)	1989-1990

Studies to Evaluate Explosion Potential Completed Studies

Sodium Nitrate (NaNO_3) (Saltcake) and Organics

- **Evaluation by HRI (1973-1977)**
 - Laboratory studies on mixtures of sodium nitrate, saltcake, and major organic chemicals used at the Hanford Site
 - Reports issued in 1976 and 1977
- **Conclusion**
 - These mixtures are stable at temperatures below 480 °F. Current maximum reading in SSTs is approximately 230 °F.

Studies to Evaluate Explosion Potential Completed Studies (cont)

Organics

- **Evaluation conducted by PNL (1985)**
 - **Literature review of organic explosives (fulminates, lead azide, phenolic compounds, polynitrated organics, nitrate esters, black powder explosives) and one inorganic explosive (ammonium nitrate)**
- **Conclusion**
 - **Operating conditions (i.e., alkaline, high moisture, low temperature) do not allow the presence of these explosives in the tanks.**

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Studies to Evaluate Explosion Potential Studies in Progress

FeCN + NaNO₃ or NaNO₂ (22 tanks)

- **Laboratory testing by PNL and Los Alamos National Laboratory (1988-1990)**
- **Lowest reaction temperature observed (460 °F) is substantially higher than the maximum temperature measured in FeCN-containing tanks (134 °F)**

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Kyshtym Explosion History

- **Waste tank explosion occurred on September 29, 1957**
 - **Contaminated large areas (primarily with ^{90}Sr)**
 - $>15,000 \text{ km}^2$ to $>0.1 \text{ Ci/km}^2$**
 - $1,000 \text{ km}^2$ to $>2 \text{ Ci/km}^2$**
 - 120 km^2 to $>100 \text{ Ci/km}^2$**
 - **10,000 people evacuated**
- **Confirmed by USSR**
 - **TASS, June 17, 1989**
 - **Technical report through the International Atomic Energy Agency, July 26, 1989**

Kyshtym Explosion History(cont)

- **First revealed by Medvedev, geneticist exiled in 1973**
- **Event analyzed by**
 - **Prof. Frank L. Parker, Vanderbilt University, in 1978**
 - **Oak Ridge National Laboratory in 1979**
 - **Los Alamos National Laboratory in 1982**

Kyshtym Explosion

Evaluation

- **Chemical Explosion**
 - Nitrate-acetate – most likely as mixed sodium saltcake
 - Cake dried by radiolytic heating months after cooling water system failure
 - Autoignition temperature ≥ 716 °F (i.e., decomposition of NaNO_3)
- Acetates probably introduced by chemical processing
- Cesium-rich supernatants apparently drawn off earlier

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Relevance of Kyshtym to Hanford Site SST Waste

Chemicals that could react are present in SST waste

- **FeCN-NaNO₃**
 - **Lowest reaction temperature observed was 460 °F**
 - **FeCN tanks (22) are currently moist and cool (<135 °F)**
- **Organics – NaNO₃**
 - **Autoignition temperature of NaNO₃ is 716 °F**
 - **Explosion potential low at organic concentration of <5% in NaNO₃**
 - **SSTs are currently moist and cool (<230 °F)**

Summary and Conclusions

- Potential for chemical explosion in SSTs has been evaluated throughout the past 15 to 20 years
- Explosion or rapid reactions from the types of chemicals present in SSTs, in general, require both specific chemical mixtures and high temperatures
- Chemicals that could react are present in tanks but high temperatures are not present in tanks
- Studies in progress to further evaluate FeCN will be completed in 1990